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equation of the following reactions a. Sodium reacts with oxygen
to produce a white metallic oxide.

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3.1 Relative Atomic Mass and Relative Molecular Mass 3.2 The Mole and the Number of Particles 3.3 The Mole and the Mass of Substances

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8.1 Understanding Salts 8.2 Preparing Salts 8.3 Qualitative Analysis – Identifying Gases 8.4 Qualitative Analysis – Colours 8.5 Qualitative Analysis – Heating Effect 8.6 Qualitative Analysis – Identifying Anion 8.7 Qualitative Analysis – Identifying Cation

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Short Notes: Form 4 Chemistry Chemical Formulae and Equation Calculation . Summary Summary \div molar volume \times molar volume \times molar mass . Avogadro Constant . Number of particles . Mole of particles . Mass of particle (in gram) ... $4 + 2\text{NH}_3 \rightarrow$

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(NH. 4) 2. SO. 4. CH. 3. COOH + KOH \rightarrow CH. 3.

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Revision Notes Redox Reaction Redox Reaction in Terms of Gain and Loss of Oxygen Redox Reaction in Terms of Gain and Loss of Hydrogen Oxidation Number Oxidation Number and IUPAC Nomenclature Redox Reaction in Terms of Change of Oxidation Number Redox Reaction in Terms of Transfer of Electron Types of Redox Reaction Oxidising and Reducing Agents Change of Iron(II) to Iron(III) and Vice Versa Displacement ...

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Martin is a well-known Chemistry specialist. 15 years of coaching experience helps him to understand the true reasons behind students who are not performing ...

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Form 5 note_chapter_3 1. FORM 5 CHAPTER 3 COORDINATION AND RESPONSE K (hinkTnkC irsten T a entre) 2. COORDINATION AND RESPONSE1. An organism experiences changes in its internal and external environments all the time2. The changes which cause responses in the body are called stimuli3.

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