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Element Carbon
Hydrogen Oxygen
Percentage 62.1 10.3
27.6. Mass in 100 g
62.1 g 10.3 g 27.6 g.
Number of moles. 12.0
 $62.1 = 5.18$ 1.0 $10.3 =$
 10.3 16.0 $27.6 = 1.73$
[1] Relative number of
atoms 1.73 $5.18 = 3$
 1.73 $10.3 = 6$ 1.73
 $1.73 = 1$.

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13 Marking scheme:
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a. $C_nH_{2n+1}OH$ [1]

cButan-1-ol and

2-methylpropan-2-ol

[1] structural isomers

[1] dCompounds with

the same molecular

formula [1] but

different

structural/displayed

formulae [1]

e2-methylbutan-2-ol

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[1] $2 \text{C}_6\text{H}_{12}\text{O}_6 \rightarrow$
 $2\text{CH}_3\text{CH}_2\text{OH} +$
 2CO_2 [1] fermentation
[1]

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5 a i Glucagon
secretion increases in

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Describe the sequence
of events that takes
place in ...

2 End-of-chapter

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parts into which a book
or other piece of text is

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end definition: 1. the
part of a place or thing
that is furthest away

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from the centre: 2. the final part of.... Learn more.

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level coursebook to
end of chapter
questions.

Answers To End Of Chapter Questions Chemistry

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International A Level
Physics Answers to end-
of-chapter questions
Answers to EOC
questions Chapter 26 b
The force on strip A is
towards strip B and the
force on strip B is
towards strip A, i.e. the

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1 a (The force F is given by $F = BIl \sin \theta$.)
strips attract each other.

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